



Synthesis of enantiopure 1,2-diamine attached to cross-linked polystyrene and its application to an insoluble catalyst for asymmetric hydrogenation

Shinichi Itsuno,* Atsushi Tsuji and Miyuki Takahashi

Department of Materials Science, Toyohashi University of Technology, Toyohashi 441-8580, Japan

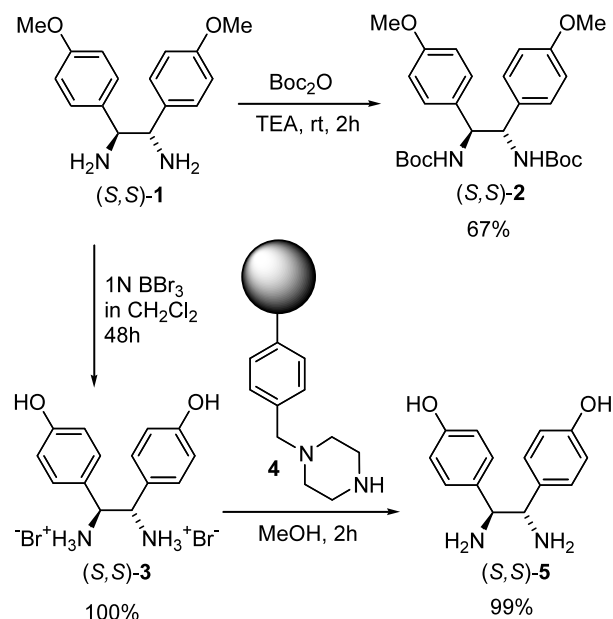
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Abstract—A novel enantiopure 1,2-diamine (**5**) having two phenolic hydroxy groups was attached into chloromethylated polystyrene through benzyl ether linkage, which was used as a chiral ligand of the catalyst in asymmetric hydrogenation of acetophenone. © 2003 Elsevier Science Ltd. All rights reserved.

Enantiopure 1,2-diamines have recently been known as an efficient chiral auxiliary in various kinds of chiral catalysts and reagents.¹ Although a variety of 1,2-diamine structures have been designed and synthesized, no attention has been paid to preparation of the 1,2-diamines possessing phenolic hydroxy groups such as **5**. The phenol groups would be a very useful functionality to attach the chiral 1,2-diamine moiety into solid support materials. Supported catalysts have found increasing use in synthesis since they have inherent operational and economical advantages.² Surprisingly, it appears that there is no report up until now on the immobilization of chiral 1,2-diamine.³ We now report the first preparation of a polymer-supported chiral 1,2-diamine **5P** by means of the reaction between **5** and chloromethylated polystyrene **6**. We also demonstrate the asymmetric hydrogenation of acetophenone as a test reaction using **5P** complexed with RuCl₂ and BINAP.⁴

According to Corey's method,⁵ racemic 1,2-diamine **1** was readily prepared from 4,4'-dimethoxybenzil as a starting material. Optical resolution of racemic **1** was successfully achieved with (L)-(+)-tartaric acid.⁶ Enantiomeric purity of the obtained chiral 1,2-diamine (*S,S*)-**1** was determined by HPLC analysis of its di(*t*-butyloxycarbonyl) derivative (*S,S*)-**2**. After optical resolution (*S,S*)-**1** was treated with BBr₃ in CH₂Cl₂ to

cleave its methyl ether linkage to yield dihydroxydiamine dihydrobromide (*S,S*)-**3**. Once the usual aqueous alkaline treatment was undertaken for (*S,S*)-**3**, extraction of the desired chiral 1,2-diamine having phenol groups resulted in failure due to its solubility problem. We found that the use of polymer-supported piperazine **4**⁷ as a scavenger resin was quite effective to remove boron derivatives and neutralize the HBr salt to afford (*S,S*)-**5** in quantitative yield (Scheme 1).



Scheme 1. Preparation of chiral 1,2-diamine **5**.

Keywords: 1,2-diamine; cross-linked polystyrene; asymmetric hydrogenation.

* Corresponding author. Tel.: +81-(0)532-44-6813; fax: +81-(0)532-44-6813; e-mail: itsuno@tutms.tut.ac.jp

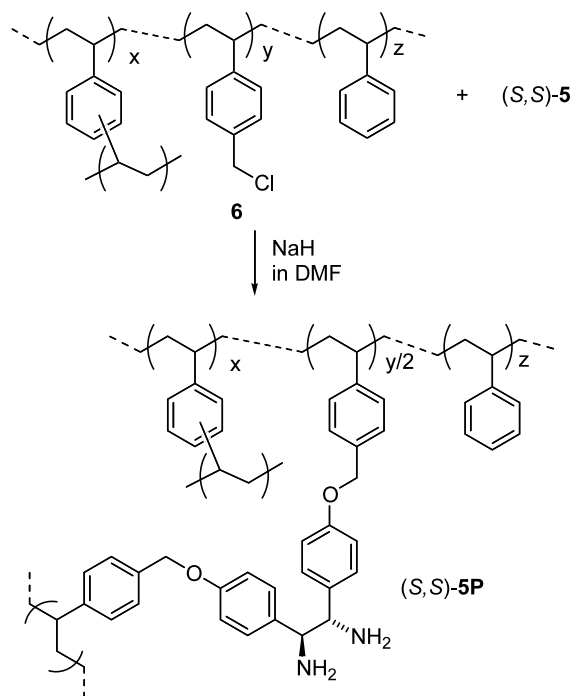
Our previous study revealed that a phenolic functionality is quite suitable for immobilization reaction to chloromethylated polystyrenes.⁸ Thus, the sodium phenoxide derived from (*S,S*)-**5** was allowed to react with 1% cross-linked chloromethylated polystyrene to give polymer-supported diamine (*S,S*)-**5P**, as shown in Scheme 2. In this immobilization reaction, both hydroxy groups of (*S,S*)-**5** seem to react with the chloromethyl groups of **6** based on the diamine loading of **5P** calculated by nitrogen analysis. Under the same reaction conditions, 1,2-diphenylethylenediamine (DPEN) was not incorporated to the polymer, which means that no nucleophilic substitution reaction between the primary amino groups of 1,2-diamine and the chloromethyl groups in **6** took place. In the cases of higher content of the chloromethyl group (**5Pg**, **5Ph**) and a higher degree of crosslinking (**5Pe**), some unreacted chloromethyl groups remained.

One of the most important asymmetric reactions using enantiopure 1,2-diamine as a chiral ligand is hydrogenation of simple ketones developed by Noyori et al.⁹ Although Noyori reported the use of a polymeric catalyst prepared from polymer-supported BINAP, RuCl₂

and DPEN for the same reaction,¹⁰ no example has been reported on the use of polymeric 1,2-diamine. In order to demonstrate the efficiency of the polymer-supported chiral 1,2-diamine **5P**, we performed the asymmetric hydrogenation of acetophenone using the polymer-supported Ru complex derived from (*S*)-BINAP, RuCl₂ and (*S,S*)-**5P**. The results are summarized in Table 1. In most cases 2-propanol is the choice of solvent for the hydrogenation using this catalyst system.⁹ Unfortunately, no reaction was achieved by using the polymer-supported complex in pure 2-propanol as a solvent (run 5). Styrene-based cross-linked polymers show little swelling in an alcoholic solvent which thus may prevent accessibility to the catalyst site in the polymer network. Addition of DMF as a co-solvent remarkably improved the reactivity of the hydrogenation using the polymeric catalyst. Acetophenone was smoothly hydrogenated with the polymer-supported Ru complex derived from (*S,S*)-**5P** and (*S*)-BINAP–RuCl₂ in a 1:1 2-propanol/DMF mixture to give the *R* alcohol with 73% ee in 100% yield (run 7).¹¹ This result strongly supported that the active complex was formed in the polymer since BINAP–RuCl₂ alone gives no reaction.¹² Pure DMF retarded the reaction with the polymeric catalyst (run 10). In a 1:1 2-propanol/DMF mixture a homogeneous catalyst derived from DPEN yielded the same alcohol with 81% ee in 92% yield (run 1). In DMF the ee decreased to 72% (run 2). Instead of BINAP the use of xylBINAP produced the same product in higher enantioselectivity as expected from the asymmetric hydrogenation data reported by Noyori (run 11).¹³ The degree of crosslinking and loading influenced both the reactivity and the enantioselectivity. A lower degree of crosslinking gave better reactivity. Higher ees and conversions were obtained when **5Pc** was used.

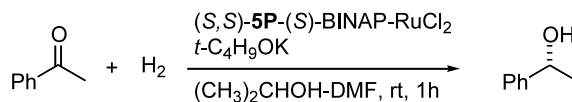
Isolation of the product by simple filtration was quite easy when the polymer-supported chiral catalyst was used. The recycling experiments were also readily performed. After a reaction was completed and the polymer-supported catalyst was allowed to settle, the solution containing the product was removed through a syringe. A further amount of acetophenone in 2-propanol/DMF was injected and the reaction was repeated. When (*R,R*)-**5Pc**–(*R*)-BINAP–RuCl₂ was employed, quantitative yields were obtained in the four continuous recycling experiments and the ee values of the chiral product were 73, 73, 73, and 74% ee, respectively.

In summary, we have shown an efficient preparative route to the polymer-supported chiral 1,2-diamine **5P**. Asymmetric hydrogenation of acetophenone smoothly occurred by using the **5P**–BINAP–RuCl₂ system in 2-propanol/DMF. We have shown that the immobilized catalyst can be reused at least four times without loss of activity. Studies of other asymmetric reactions using the polymeric chiral 1,2-diamine are currently under way.



(<i>S,S</i>)- 5P	x	y	z
(<i>S,S</i>)- 5Pa	0.01	0.05	0.94
(<i>S,S</i>)- 5Pb	0.01	0.10	0.89
(<i>S,S</i>)- 5Pc	0.01	0.20	0.79
(<i>S,S</i>)- 5Pd	0.02	0.20	0.78
(<i>S,S</i>)- 5Pe	0.05	0.20	0.75
(<i>S,S</i>)- 5Pf	0.01	0.30	0.69
(<i>S,S</i>)- 5Pg	0.01	0.40	0.59
(<i>S,S</i>)- 5Ph	0.01	0.50	0.49

Scheme 2. Preparation of polymer-supported chiral 1,2-diamine.

Table 1. Asymmetric hydrogenation of acetophenone using the (*S,S*)-**5P**–(*S*)-BINAP–RuCl₂ complex^a

Run	(<i>S,S</i>)- 5P	Solvent (2-propanol:DMF)	1-Phenylethanol		
			Yield (%) ^b	Ee (%) ^c	Config.
1	(<i>R,R</i>)-DPEN ^d	1:1	92	81	<i>S</i>
2	(<i>R,R</i>)-DPEN ^d	DMF	100	72	<i>S</i>
3	(<i>S,S</i>)- 5Pa	1:1	87	73	<i>R</i>
4	(<i>S,S</i>)- 5Pb	1:1	86	74	<i>R</i>
5	(<i>S,S</i>)- 5Pc	2-Propanol	0	–	–
6	(<i>S,S</i>)- 5Pc	2:1	46	71	<i>R</i>
7	(<i>S,S</i>)- 5Pc	1:1	100	73	<i>R</i>
8	(<i>R,R</i>)- 5Pc ^d	1:1	100	73	<i>S</i>
9	(<i>S,S</i>)- 5Pc	1:2	100	70	<i>R</i>
10	(<i>S,S</i>)- 5Pc	DMF	38	75	<i>R</i>
11	(<i>S,S</i>)- 5Pc ^e	1:1	100	93	<i>R</i>
12	(<i>S,S</i>)- 5Pd	1:1	69	73	<i>R</i>
13	(<i>S,S</i>)- 5Pe ^f	1:1	39	68	<i>R</i>
14	(<i>S,S</i>)- 5Pf	1:1	91	69	<i>R</i>
15	(<i>S,S</i>)- 5Pg ^f	1:1	78	67	<i>R</i>
16	(<i>S,S</i>)- 5Ph ^f	1:1	43	63	<i>R</i>

^a Reactions were conducted at 1 MPa of H₂ and at room temperature for 1 h using acetophenone (5 mmol), *t*-BuOK (1 M *t*-BuOH soln. 100 μL), 1,2-diamine (0.025 mmol) and BINAP–RuCl₂ (0.025 mmol).

^b Determined by GC.

^c Determined by HPLC using Chiralcel OD.

^d (*R*)-BINAP was used.

^e (*S*)-xylBINAP (2,2'-bis(di-3,5-xylylphosphino)-1,1'-binaphthyl) was used.

^f Contains unreacted chloromethyl groups.

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propanol solution of *t*-BuOK (100 μ L, 0.1 mmol) in 2-propanol (2 mL), and then, hydrogen was pressurized to 1 MPa. The suspension was stirred at room temperature for 1 h. After the reaction, the mixture was filtered and concentrated. The yield determined by GC was 100%. The enantioselectivity was determined by HPLC analysis using a Daicel Chiralcel OD column (eluent, 1:20 2-propanol–hexane; flow rate, 0.4 mL/min).

12. After the complex formation of the polymer-supported 1,2-diamine with BINAP, some amount of free BINAP–RuCl₂ was detected in the solution part by NMR analysis. Based on the NMR analysis of the solution part, we can estimate that at least 85% of the polymeric 1,2-diamine forms the complex with BINAP–RuCl₂.
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